SET - 3

Series : SSO/C

कोड नं. Code No.

55/3

रोल नं.				
Roll No.				

परीक्षार्थी कोड को उत्तर-पुस्तिका के मुख-पृष्ठ पर अवश्य लिखें ।

Candidates must write the Code on the title page of the answer-book.

- कृपया जाँच कर लें कि इस प्रश्न-पत्र में मुद्रित पृष्ठ 12 हैं ।
- प्रश्न-पत्र में दाहिने हाथ की ओर दिए गए कोड नम्बर को छात्र उत्तर-पुस्तिका के मुख-पृष्ठ पर लिखें ।
- कुपया जाँच कर लें कि इस प्रश्न-पत्र में 26 प्रश्न हैं ।
- कृपया प्रश्न का उत्तर लिखना शुरू करने से पहले, प्रश्न का क्रमांक अवश्य लिखें।
- इस प्रश्न-पत्र को पढ़ने के लिए 15 मिनट का समय दिया गया है । प्रश्न-पत्र का वितरण पूर्वाह्न में 10.15 बजे किया जायेगा । 10.15 बजे से 10.30 बजे तक छात्र केवल प्रश्न-पत्र को पढ़ेंगे और इस अवधि के दौरान वे उत्तर-पुस्तिका पर कोई उत्तर नहीं लिखेंगे ।
- Please check that this question paper contains 12 printed pages.
- Code number given on the right hand side of the question paper should be written on the title page of the answer-book by the candidate.
- Please check that this question paper contains **26** questions.
- Please write down the Serial Number of the question before attempting it.
- 15 minutes time has been allotted to read this question paper. The question paper will be distributed at 10.15 a.m. From 10.15 a.m. to 10.30 a.m., the students will read the question paper only and will not write any answer on the answer-book during this period.

भौतिक विज्ञान (सैद्धान्तिक) PHYSICS (Theory)

निर्धारित समय :3 घंटे] [अधिकतम अंक :70

Time allowed: 3 hours] [Maximum Marks: 70

सामान्य निर्देश:

- (i) **सभी** प्रश्न अनिवार्य हैं । इस प्रश्न-पत्र में कुल 26 प्रश्न हैं ।
- (ii) इस प्रश्न-पत्र के **5** भाग हैं : खण्ड-**अ**, खण्ड-**ब**, खण्ड-**स**, खण्ड-**द** और खण्ड-**य** ।
- (iii) खण्ड-अ में 5 प्रश्न हैं, प्रत्येक का 1 अंक है । खण्ड-ब में 5 प्रश्न हैं, प्रत्येक के 2 अंक हैं । खण्ड-स में 12 प्रश्न हैं, प्रत्येक के 3 अंक हैं । खण्ड-द में 4 अंक का एक मूल्याधारित प्रश्न है और खण्ड-य में 3 प्रश्न हैं, प्रत्येक के 5 अंक हैं ।
- (iv) प्रश्न-पत्र में समग्र पर कोई विकल्प नहीं है । तथापि, **दो** अंकों वाले **एक** प्रश्न में, **तीन** अंकों वाले **एक** प्रश्न में और **पाँच** अंकों वाले **तीनों** प्रश्नों में आन्तरिक चयन प्रदान किया गया है । ऐसे प्रश्नों में आपको दिए गए चयन में से केवल **एक** प्रश्न ही करना है ।

55/3 1 [P.T.O.



(v) जहाँ आवश्यक हो आप निम्नलिखित भौतिक नियतांकों के मानों का उपयोग कर सकते हैं :

$$c=3\times10^8$$
 m/s $h=6.63\times10^{-34}$ Js $e=1.6\times10^{-19}$ C $\mu_0=4\pi\times10^{-7}$ T m A^{-1} $\epsilon_0=8.854\times10^{-12}$ C^2 N^{-1} m^{-2} $\frac{1}{4\pi\epsilon_0}=9\times10^9$ N m 2 C $^{-2}$ $m_e=9.1\times10^{-31}$ kg न्यूट्रॉन का द्रव्यमान $=1.675\times10^{-27}$ kg आवोगाद्रो संख्या $=6.023\times10^{23}$ प्रति ग्राम मोल बोल्टजमान नियतांक $=1.38\times10^{-23}$ JK $^{-1}$

General Instructions:

- (i) All questions are compulsory. There are 26 questions in all.
- (ii) This question paper has **five** sections: Section **A**, Section **B**, Section **C**, Section **D** and Section **E**.
- (iii) Section A contains five questions of one mark each, Section B contains five questions of two marks each, Section C contains twelve questions of three marks each, Section D contains one value based question of four marks and Section E contains three questions of five marks each.
- (iv) There is no overall choice. However, an internal choice has been provided in **one** question of **two** marks, **one** question of **three** marks and all the **three** questions of **five** marks weightage. You have to attempt only **one** of the choices in such questions.
- (v) You may use the following values of physical constants wherever necessary:

$$c = 3 \times 10^8 \text{ m/s}$$

$$h = 6.63 \times 10^{-34} \text{ Js}$$

$$e = 1.6 \times 10^{-19} \text{ C}$$

$$\mu_0 = 4\pi \times 10^{-7} \text{ T m A}^{-1}$$

$$\epsilon_0 = 8.854 \times 10^{-12} \text{ C}^2 \text{ N}^{-1} \text{ m}^{-2}$$

$$\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$$

$$m_e = 9.1 \times 10^{-31} \text{ kg}$$
mass of neutron = 1.675 × 10⁻²⁷ kg
mass of proton = 1.673 × 10⁻²⁷ kg
Avogadro's number = 6.023 × 10²³ per gram mole
Boltzmann constant = 1.38 × 10⁻²³ JK⁻¹

55/3



खण्ड – अ

Section - A

- 1. NAND गेट का तर्क प्रतीक खींचिए और इसकी सत्यमान सारणी दीजिए ।

 Draw the logic symbol of NAND gate and give its Truth Table.
- 2. AC स्रोत की आवृत्ति में परिवर्तन के साथ संधारित्र प्रतिघात में विचरण को दर्शाने के लिए ग्राफ खींचिए।
 Plot a graph showing variation of capacitive reactance with the change in the frequency of the AC source.
- 3. आयाम मॉडुलन और आवृत्ति मॉडुलन के बीच विभेदन कीजिए ।

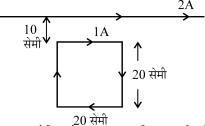
 Distinguish between amplitude modulation and frequency modulation.
- 4. किसी चालक के समिवभव पृष्ठ के किसी बिन्दु पर विद्युत क्षेत्र रेखाएँ पृष्ठ के लम्बवत क्यों होती हैं ?

 Why are electric field lines perpendicular at a point on an equipotential surface of a conductor ?
- 5. किसी संधारित्र को किसी परिवर्ती आवृत्ति के AC स्रोत से संयोजित किया गया है। यदि AC स्रोत की आवृत्ति घटा दी जाए, तो क्या विस्थापन धारा परिवर्तित हो जाएगी?

 A variable frequency AC source is connected to a capacitor. Will the displacement current change if the frequency of the AC source is decreased?

खण्ड – ब Section – B

6. 20 cm भुजा वाले किसी वर्गाकार लूप जिससे 1A धारा प्रवाहित हो रही है, को किसी अनन्त लम्बाई के सीधे तार जिससे 2A धारा प्रवाहित हो रही है के निकट चित्र में दर्शाए अनुसार समान तल में रखा गया है ।

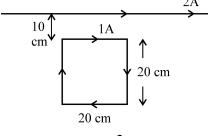


धारावाही चालक के कारण लूप पर आरोपित नेंट बल का परिमाण और दिशा परिकलित कीजिए ।

अथवा

200 फेरों और $100~{
m cm}^2$ क्षेत्रफल की किसी वर्गाकार समतल कुण्डली से $5{
m A}$ अपरिवर्ती धारा प्रवाहित हो रही है । यह कुण्डली $0.2~{
m T}$ के ऐसे एकसमान चुम्बकीय क्षेत्र में स्थित है, जिसकी दिशा कुण्डली के तल के लम्बवत है । जब इस कुण्डली का तल चुम्बकीय क्षेत्र से 60° का कोण बनाता है तब उस स्थिति में कुण्डली पर लगा बल-आघूर्ण परिकलित कीजिए । किस विन्यास में यह कुण्डली स्थायी साम्यावस्था में होगी ?

A square loop of side 20 cm carrying current of 1A is kept near an infinite long straight wire carrying a current of 2A in the same plane as shown in the figure.



55/3 3 [P.T.O.

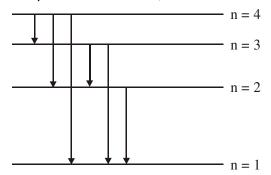
Calculate the magnitude and direction of the net force exerted on the loop due to the current carrying conductor.

OR

A square shaped plane coil of area 100 cm² of 200 turns carries a steady current of 5A. It is placed in a uniform magnetic field of 0.2 T acting perpendicular to the plane of the coil. Calculate the torque on the coil when its plane makes an angle of 60° with the direction of the field. In which orientation will the coil be in stable equilibrium?

- उन वैद्युत चुम्बकीय विकिरणों का नाम लिखिए (i) जिनका उपयोग कैंसर की कोशिकाओं को नष्ट करने में 7. किया जाता है, (ii) जिनसे त्वचा ताम्र रंग की हो जाती है, (iii) पृथ्वी की उष्णता बनाए रखते हैं। 2 इनमें से किसी एक प्रकार की तरंगों को उत्पन्न करने की विधि का संक्षेप में वर्णन कीजिए । Name the types of e.m. radiations which (i) are used in destroying cancer cells, (ii) cause tanning of the skin and (iii) maintain the earth's warmth. Write briefly a method of producing any one of these waves.
- चित्र में हाइड्रोजन परमाणु का ऊर्जा स्तर आरेख दर्शाया गया है : 8.

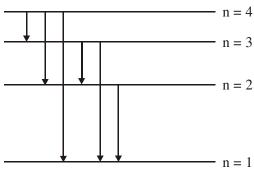
वह संक्रमण ज्ञात कीजिए जिसमें 496 nm तरंगदैर्ध्य के फोटॉन का उत्सर्जन होता है ।



किस संक्रमण के संगत अधिकतम तरंगदैध्यं के विकिरण उत्सर्जित होते हैं ? अपने उत्तर की पृष्टि (b)

The figure shows energy level diagram of hydrogen atom.

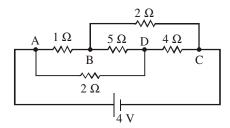
Find out the transition which results in the emission of a photon of wavelength (a) 496 nm.



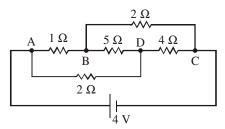
(b) Which transition corresponds to the emission of radiation of maximum wavelength? Justify your answer.

55/3

- 9. 'विद्युत फ्लक्स' की परिभाषा और इसका SI मात्रक लिखिए । विद्युत क्षेत्र $\overrightarrow{E} = 3 \times 10^3 \, \hat{i} \, \text{N/C}$ के कारण किसी $10 \, \text{cm}$ भुजा वाले वर्ग से गुजरने वाला फ्लक्स कितना है, जबिक इसे \overrightarrow{E} के अभिलम्बवत् रखा गया है । 2 Define the term 'electric flux'. Write its SI units. What is the flux due to electric field $\overrightarrow{E} = 3 \times 10^3 \, \hat{i} \, \text{N/C}$ through a square of side $10 \, \text{cm}$, when it is held normal to \overrightarrow{E} ?
- 10. चित्र में दर्शाए गए प्रतिरोधकों के नेटवर्क द्वारा बैटरी से ली गयी धारा परिकलित कीजिए ।



Calculate the current drawn from the battery by the network of resistors shown in the figure.



खण्ड – स

Section - C

- 11. (a) साधारण रंगीन काँच के चश्मों की तुलना में अच्छी गुणता के पोलरॉयडों के बने चश्मों को प्रायिकता क्यों दी जाती है ? कारण देकर स्पष्ट कीजिए ।
 - (b) दो पोलरॉयडों P_1 तथा P_2 को क्रांसित स्थितियों में रखा गया है । P_1 और P_2 के बीच कोई तीसरा पोलरॉयड P_3 इस प्रकार रखा जाता है कि P_3 का पारित अक्ष P_1 के समान्तर है । P_2 से पारगिमत प्रकाश की तीव्रता (I_2) P_3 को घूर्णन कराने पर किस प्रकार परिवर्तन होगी ? P_1 और P_3 के परित अक्षों के बीच कोण ' θ ' और तीव्रता ' I_2 ' के बीच ग्राफ खींचिए ।
 - (a) Good quality sun-glasses made of polaroids are preferred over ordinary coloured glasses. Justifying your answer.
 - (b) Two polaroids P_1 and P_2 are placed in crossed positions. A third polaroid P_3 is kept between P_1 and P_2 such that pass axis of P_3 is parallel to that of P_1 . How would the intensity of light (I_2) transmitted through P_2 vary as P_3 is rotated? Draw a plot of intensity I_2 Vs the angle I_3 , between pass axes of I_3 and I_3 .

55/3 5 [P.T.O.



- 12. (a) नीचे दी गयी नाभिकीय अभिक्रियाओं को पूरा कीजिए:
 - (i) ${}^{208}_{84}\text{Po} \rightarrow {}^{204}_{82}\text{Pb} + \dots$
 - (ii) ${}^{32}_{15}P \rightarrow {}^{32}_{16}S + ...$
 - (b) (i) β^- और (ii) β^+ क्षय के लिए उत्तरदायी नाभिक में अन्तर्ग्रस्त मूल प्रक्रिया लिखिए ।
 - (c) न्यूट्रिनों का प्रायोगिक संसूचन कठिन क्यों पाया गया ?
 - (a) Complete the following nuclear reactions:
 - (i) ${}^{208}_{84}\text{Po} \rightarrow {}^{204}_{82}\text{Pb} + \dots$
 - (ii) ${}^{32}_{15}P \rightarrow {}^{32}_{16}S + ...$
 - (b) Write the basic process involved in nuclei responsible for (i) β^- and (ii) β^+ decay.
 - (c) Why is it found experimentally difficult to detect neutrinos?
- 13. ताप T > 0K पर (i) n-प्रकार और (ii) p-प्रकार के अर्धचालकों के ऊर्जा-बैण्ड आरेख खींचिए । n-प्रकार के प्रकरण में Si-अर्धचालकों के ऊर्जा-बैण्ड आरेख में दाता ऊर्जा स्तर चालक बैण्ड की तली के कुछ नीचे तथा p-प्रकार के अर्धचालकों में ग्राही ऊर्जा स्तर संयोजी बैण्ड के शीर्ष से कुछ ऊपर होता है । स्पष्ट कीजिए कि चालक और संयोजी बैण्डों में इन ऊर्जा-स्तरों की क्या भूमिका होती है ।

Draw the energy band diagrams of (i) n-type and (ii) p-type semiconductor at temperature, T > 0K.

In the case n-type Si semiconductor, the donor energy level is slightly below the bottom of conduction band whereas in p-type semiconductor the acceptor energy level is slightly above the top of the valence band. Explain, what role do these energy levels play in conduction and valence bands.

14. अंतरण अभिलक्षण (V_0 और V_i के बीच) का ग्राफ खींचिए और दर्शाइए कि इस अभिलक्षण का कौन सा भाग प्रवर्धन के लिए उपयोग किया जाता है और क्यों ?

किसी आधार बायिसत ट्रांजिस्टर प्रवर्धक का CE विन्यास में परिपथ आरेख खींचिए और संक्षेप में इसकी कार्यविधि की व्याख्या कीजिए ।

Draw a plot of transfer characteristic $(V_0 \text{ vs } V_i)$ and show which portion of the characteristic is used in amplification and why?

Draw the circuit diagram of base bias transistor amplifier in CE configuration and briefly explain its working.

- 15. इन्टरनेट के उपयोग में नीचे दिए गए पदों की व्याख्या कीजिए :
 - (i) इन्टरनेट सर्फिन्ग
 - (ii) सोशल नेटवर्किंग
 - (iii) ई-मेल

Explain the following terms in relation to the use of internet:

- (i) Internet surfing
- (ii) Social networking
- (iii) E-mail

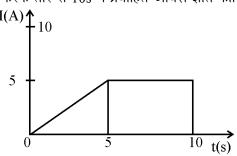
55/3



3

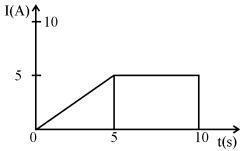
3

3



(a) Deduce the relation between current I flowing through a conductor and drift velocity $\overrightarrow{V_d}$ of the electrons.

(b) Figure shows a plot of current 'I' flowing through the cross-section of a wire versus the time 't'. Use the plot to find the charge flowing in 10s through the wire.



17. किसी पोटैन्शियोमीटर का परिपथ आरेख खींचकर इसका कार्यकारी सिद्धान्त लिखिए । दो सेलों की emf की तुलना करने में इसका उपयोग किस प्रकार किया जाता है इसका वर्णन करने वाला आवश्यक सूत्र व्युत्पन्न कीजिए ।

अथवा

परिपथ आरेख की सहायता से मीटर सेतु के कार्यकारी सिद्धान्त की व्याख्या कीजिए । इसका उपयोग किसी दिए गए तार का अज्ञात प्रतिरोध निर्धारित करने में किस प्रकार किया जाता है ? परिणाम में न्यूनतम त्रुटि के लिए बरती जाने वाली आवश्यक सावधानियाँ लिखिए ।

Draw a circuit diagram of a potentiometer. State its working principle. Derive the necessary formula to describe how it is used to compare the emfs of the two cells.

OR

With the help of the circuit diagram, explain the working principle of meter bridge. How is it used to determine the unknown resistance of a given wire? Write the necessary precautions to minimize the error in the result.

18. (a) किसी चल कुण्डली गैल्वेनोमीटर में चुम्बकीय क्षेत्र त्रिज्य (अरीय) क्यों बनाया जाता है ? समझाइए यह किस प्रकार बनाया जाता है ।

(b) किसी गैल्वेनोमीटर जिसका प्रतिरोध 'G' है, को इसके श्रेणी में कोई प्रतिरोध 'R' संयोजित करके (0-V) वोल्ट परिसर के वोल्टमीटर में परिवर्तित किया जा सकता है । यदि इसे 0 से V/2 परिसर के वोल्टमीटर में परिवर्तित करना हो, तो कितने प्रतिरोध की आवश्यकता होगी ?

55/3 7 [P.T.O.



3

- (a) Why is the magnetic field radial in a moving coil galvanometer? Explain how it is achieved.
- A galvanometer of resistance 'G' can be converted into a voltmeter of range (b) (0-V) volts by connecting a resistance 'R' in series with it. How much resistance will be required to change its range from 0 to V/2?
- किसी ac वोल्टता $V = V_0 \sin \omega t$ के स्रोत को प्रतिरोध 'R' और संधारित्र 'C' के श्रेणी संयोजन से जोड़ा 19. गया है । इसके लिए फेज़र आरेख खींचिए और इसका उपयोग (i) परिपथ की प्रतिबाधा और (ii) कला-कोण के लिए व्यंजक प्राप्त करने में कीजिए ।

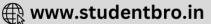
A source of ac voltage $V = V_0 \sin \omega t$ is connected to a series combination of a resistor 'R' and a capacitor 'C'. Draw the phasor diagram and use it to obtain the expression for (i) impedance of the circuit and (ii) phase angle.

अनुप्रस्थ काट-क्षेत्रफल $1.6 \times 10^{-4} \text{m}^2$ और कसकर पास-पास लपेटे गए 2000 फेरों की परिनालिका जिससे 20. 4.0 A धारा प्रवाहित हो रही है इसके केन्द्र से होकर निलम्बित है और यह क्षैतिज तल में घुम सकती है । (i) इस परिनालिका से संबद्ध चुम्बकीय आधुर्ण, (ii) यदि परिनालिका के अक्ष से 30° कोण पर कोई $7.5 \times 10^{-2} \, \mathrm{T}$ का क्षैतिज चम्बकीय क्षेत्र व्यवस्थित किया गया है, तो परिनालिका पर लगे बल-आघर्ण का परिमाण और दिशा ज्ञात कीजिए ।

A closely wound solenoid of 2000 turns and cross sectional area 1.6×10^{-4} m² carrying a current of 4.0 A is suspended through its centre allowing it to turn in a horizontal plane. Find (i) the magnetic moment associated with the solenoid, (ii) magnitude and direction of the torque on the solenoid if a horizontal magnetic field of 7.5×10^{-2} T is set up at an angle of 30° with the axis of the solenoid.

- यंग के द्विझिरी प्रयोग में दो झिरियों की मोटाई का अनुपात 4:1 है । व्यतिकरण पैटर्न में उच्चिष्ठ और 21. (a) निम्निष्ठ की तीव्रताओं के अनुपात का मुल्यांकन कीजिए ।
 - क्या व्यतिकरण पैटर्न ने चमकीली और काली फ्रिन्जें दिखाई देना. किसी भी रूप में. ऊर्जा संरक्षण नियम (b) का उल्लंघन है ? व्याख्या कीजिए ।
 - The ratio of the widths of two slits in Young's double slit experiment is 4:1. (a) Evaluate the ratio of intensities at maxima and minima in the interference pattern.
 - Does the appearance of bright and dark fringes in the interference pattern (b) violate, in any way, conservation of energy? Explain.
- उन कारकों को लिखिए जिनके द्वारा किसी दूरदर्शक की विभेदन क्षमता में वृद्धि की जा सकती है । 3 22. (a)
 - 1 mm चौडाई की एकल झिरी पर जब झिरी के अभिलम्बवत 600 nm तरंगदैर्ध्य का प्रकाश आपतन (b) करता है, तो इसके कारण प्राप्त विवर्तन पैटर्न के प्रथम कोटि उच्चिष्ठ और तृतीय कोटि निम्निष्ठ के बीच कोणीय पृथक्कन का आकलन कीजिए ।
 - Write the factors by which the resolving power of a telescope can be increased. (a)
 - Estimate the angular separation between first order maximum and third order (b) minimum of the diffraction pattern due to a single slit of width 1 mm, when light of wavelength 600 nm is incident normal on it.

55/3 8



3

3

Section - D

- विद्यालय की छुट्टी के तुरन्त बाद जैसे ही बिमला अपनी सहेलियों के साथ बाहर निकली उसने देखा कि 23. अचानक बादलों की गर्जन के साथ बिजली चमकने लगी है । उन्हें शरण के लिए कोई उपयुक्त स्थान नहीं मिल पाया । डॉ. कपूर जो वहीं से अपनी कार से गुजर रहे थे उन्होंने इन बच्चों को देखा और उन्हें अपनी कार में बैठने का प्रस्ताव दिया, यही नहीं उन्होंने इन बच्चों को इनके घरों के निकट की बस्ती में छोड़ा । बिमला के माता-पिता जो बिमला का इन्तजार कर रहे थे, यह देखा और उन्होंने डॉ. कपूर का आभार व्यक्त किया ।
 - डॉ. कपूर और बिमला के माता-पिता द्वारा किन मूल्यों को दर्शाया गया ? (1)
 - (2) विशेषकर तड़ित और गर्जन के समय कार के भीतर होना सुरक्षित क्यों माना जाता है ?
 - "परावैद्युत सामर्थ्य" पद की परिभाषा लिखिए । यह क्या सुचित करती है ? (3)

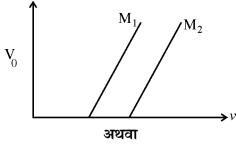
Immediately after school hour, as Bimla with her friends came out, they noticed that there was a sudden thunderstorm accompanied by the lightening. They could not find any suitable place for shelter. Dr. Kapoor who was passing thereby in his car noticed these children and offered them to come in their car. He even took care to drop them to the locality where they were staying. Bimla's parents, who were waiting, saw this and expressed their gratitude to Dr. Kapoor.

- (1) What values did Dr. Kapoor and Bimla's parents displayed?
- (2) Why is it considered safe to be inside a car especially during lightening and thunderstorm?
- Define the term 'dielectric strength'. What does this term signify? (3)

खण्ड – य

Section – E

- प्रकाश-विद्युत प्रभाव के उन तीन प्रेक्षित अभिलक्षणों को लिखिए जिनकी व्याख्या प्रकाश के तरंग 24. (a) सिद्धान्त द्वारा नहीं की जा सकती ।
 - स्पष्ट कीजिए कि आइंस्टीन के प्रकाश-विद्युत समीकरण का उपयोग इनका संतोषजनक वर्णन करने के लिए किस प्रकार किया जाता है ।
 - चित्र में दो प्रकाश-सुग्राही पदार्थों \mathbf{M}_1 तथा \mathbf{M}_2 के लिए आपितत विकिरणों की आवृत्ति (v) और निरोधी विभव (V_0) के बीच ग्राफ दर्शाया गया है । स्पष्ट कीजिए कि
 - दोनों रेखाओं का ढलान समान क्यों है ? (i)
 - आपितत विकिरणों की समान आवृत्ति के लिए किस पदार्थ से उत्सर्जित इलेक्ट्रॉनों की गतिज (ii) ऊर्जा अधिक है ?



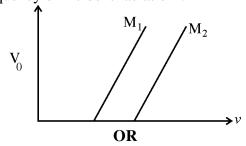
9 [P.T.O. 55/3

4

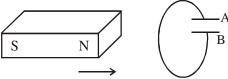
5

Get More Learning Materials Here:

- रदरफोर्ड के प्रकीर्णन प्रयोग में लक्ष्य-नाभिक के कूलॉमी क्षेत्र में lpha-कणों द्वारा अनुरेखित प्रक्षेप-पथ (a) र्खींचिए और स्पष्ट कीजिए कि किस प्रकार इसके द्वारा नाभिक के साइज़ का अनुमान लगाया गया ।
- संक्षेप में वर्णन कीजिए कि गतिमान इलेक्ट्रॉनों की तरंग प्रकृति प्रायोगिक रूप से किस प्रकार प्रमाणित (b) की गयी ।
- जब डय्टेरॉनों और α -कणों को उनकी विराम अवस्थाओं से समान त्वरित विभव. V के द्वारा त्वरित (c) किया जाता है, तो इनसे संबद्ध दे ब्राग्ली तरंगदैध्यों के अनुपात का आकलन कीजिए ।
- Write three observed features of photoelectric effect which cannot be explained (a) by wave theory of light. Explain how Einstein's photoelectric equation is used to describe these features satisfactorily.
- Figure shows a plot of stopping potential (V_0) with frequency (v) of incident (b) radiation for two photosensitive materials M_1 and M_2 . Explain
 - why the slope of both the lines is same? (i)
 - for which material emitted electrons have greater kinetic energy for the (ii) same frequency of incident radiation?



- In Rutherford scattering experiment, draw the trajectory traced by α -particles in (a) the coulomb field of target nucleus and explain how this led to estimate the size of the nucleus.
- Describe briefly how wave nature of moving electrons was established (b) experimentally.
- Estimate the ratio of de-Broglie wavelengths associated with deuterons and (c) α-particles when they are accelerated from rest through the same accelerating potential V.
- लेंज का नियम लिखिए । नीचे दी गयी स्थिति में संधारित्र की ध्रुवणता का अनुमान लगाने में इस नियम 25. (a) का उपयोग कीजिए ।



- कोई जेटयान 1800 km/h की चाल से पश्चिम दिशा में गमन कर रहा है । (b)
 - यदि जिस जगह वह गमन कर रहा है वहाँ पृथ्वी के चुम्बकीय क्षेत्र का परिमाण $5 \times 10^{-4}\,\mathrm{T}$ और नित (i) कोण 30° है, तो 25 m विस्तार की जेटयान की पंखुड़ियों के दो सिरों के बीच उत्पन्न वोल्टता-अन्तर का अनुमान लगाइए ।
 - यदि जेटयान की दिशा पश्चिम से उत्तर की ओर हो जाए. तो उत्पन्न वोल्टता पर क्या प्रभाव पड़ेगा ?

अथवा

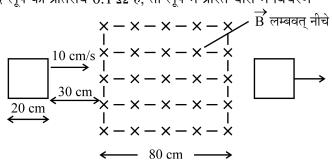
55/3 10



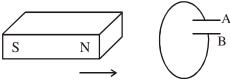
कृण्डलियों के युगल के अन्योन्य प्रेरकत्व की परिभाषा लिखिए और जिन कारकों पर यह निर्भर करता है उनका उल्लेख कीजिए ।

चित्र में दर्शाए अनुसार 20 cm भुजा का कोई वर्ग-लूप 0.1 T के एकसमान चुम्बकीय क्षेत्र के प्रभाव क्षेत्र से प्रारम्भ में 30 cm दूरी पर है । इसके पश्चात इसे दायीं ओर $10~{
m cm~s^{-1}}$ के वेग से उस समय तक गमन कराया जाता है, जब तक कि यह इस क्षेत्र के प्रभाव क्षेत्र से बाहर नहीं निकल जाता । नीचे दिए गए विचरणों को दर्शाने के लिए ग्राफ खींचिए :

- समय (t) के साथ लूप से गुजरने वाले चुम्बकीय फ्लक्स (ф) का विचरण (i)
- (ii) समय (t) के साथ लूप में प्रेरित emf (ε)
- (iii) यदि लूप का प्रतिरोध 0.1Ω है, तो लूप में प्रेरित धारा में विचरण



State Lenz's law. Use it to predict the polarity of the capacitor in the situation (a) given below:



- A jet plane is travelling towards west at a speed of 1800 km/h. (b)
 - Estimate voltage difference developed between the ends of the wing having a span of 25 m if the Earth's magnetic field at the location has a 5×10^{-4} T and dip angle is 30°.
 - How will the voltage developed be affected if the jet changes its direction from west to north?

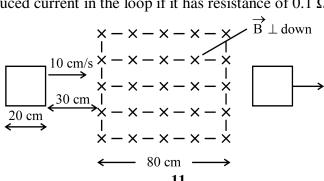
OR

Define mutual inductance of a pair of coils and write on which factors does it

A square loop of side 20 cm is initially kept 30 cm away from a region of uniform magnetic field of 0.1 T as shown in the figure. It is then moved towards the right with a velocity of 10 cm s^{-1} till it goes out of the field.

Plot a graph showing the variation of

- magnetic flux (ϕ) through the loop with time (t).
- induced emf (ε) in the loop with time t.
- (iii) induced current in the loop if it has resistance of 0.1 Ω .



[P.T.O.



- तरंगाग्र की परिभाषा लिखिए । यह किरण से किस प्रकार भिन्न है ? 26.
 - नीचे दिए गए प्रत्येक प्रकरण के लिए तरंगाग्र की आकृति दर्शाइए । (b)
 - किसी बिन्दु स्रोत से प्रकाश का अपसारित होना । (i)
 - जब बिन्दु स्रोत किसी उत्तल लेंस के फोकस पर है, तब लेंस से प्रकाश निर्गत होते हुए (ii)
 - (iii) हाइगेन्स के द्वितीयक तरंगिका के निर्माण का उपयोग करते हुए, सघन माध्यम से विरल माध्यम में किसी समतल तरंगाग्र के गमन को दर्शाने के लिए आरेख खींचिए ।

- किसी संयुक्त सुक्ष्मदर्शी द्वारा प्रतिबिम्ब बनना दर्शाने के लिए किरण आरेख खींचिए । उस स्थिति में (a) कुल आवर्धन के लिए व्यंजक प्राप्त कीजिए जब प्रतिबिम्ब अनन्त पर बनता है ।
- किसी संयुक्त सूक्ष्मदर्शी की विभेदन क्षमता किस प्रकार प्रभावित होती है, जब (b)
 - अभिदृश्यक की फोकस दूरी घट जाती है। (i)
 - प्रकाश की तरंगदैर्ध्य अधिक हो जाती है ? (ii) अपने उत्तर की पृष्टि कारण सहित कीजिए ।
- Define a wavefront. How is it different from a ray? (a)
- (b) Depict the shape of a wavefront in each of the following cases.
 - Light diverging from point source.
 - (ii) Light emerging out of a convex lens when a point source is placed at its focus.
 - (iii) Using Huygen's construction of secondary wavelets, draw a diagram showing the passage of a plane wavefront from a denser into a rarer medium.

OR

- Draw a ray diagram showing the image formation by a compound microscope. (a) Obtain expression for total magnification when the image is formed at infinity.
- How does the resolving power of a compound microscope get affected, when (b)
 - focal length of the objective is decreased.
 - the wavelength of light is increased? (ii) Give reasons to justify your answer.

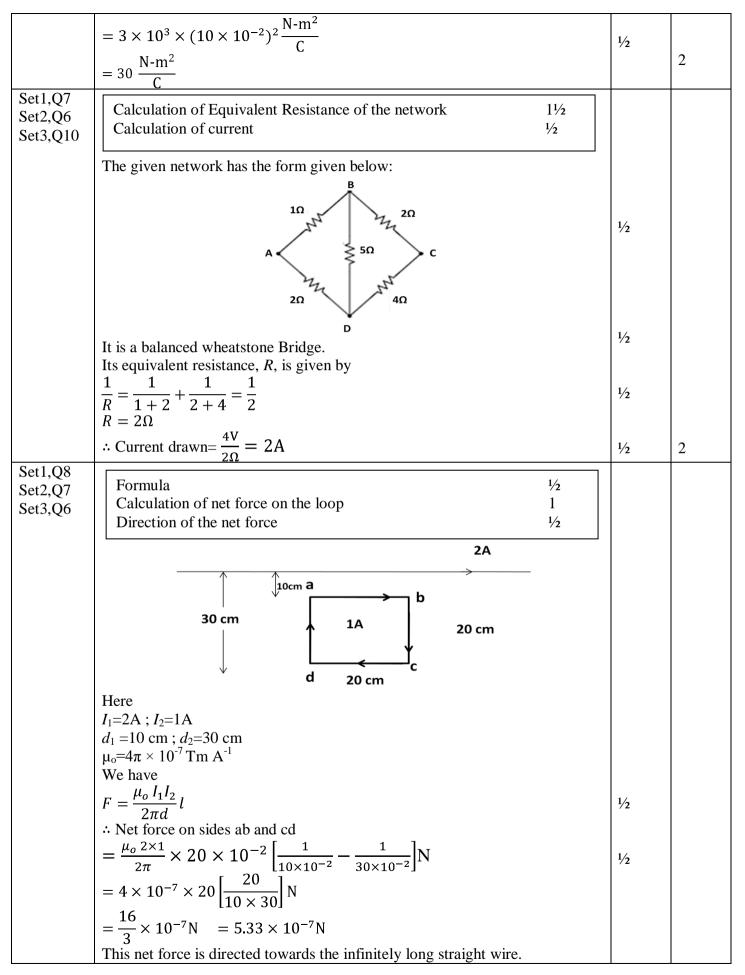
55/3 12



MARKING SCHEME SET 55/1(Compartment)

O N	SET 55/I(Compartment)	3.6	7D ()
Q. No.	Expected Answer / Value Points	Marks	Total Marks
	Section A		
Set1,Q1	If it were not so, the presence of a component of the field along the surface	1	
Set1,Q1 Set2,Q5	would violate its equipotential nature.	1	
Set2,Q3 Set3,Q4	[Accept any other correct explanation]		1
Set3,Q4 Set1,Q2	It would decrease.	1	1
Set1,Q2 Set2,Q1	[NOTE: Also accept if the student just writes 'yes']	1	
Set2,Q1 Set3,Q5	[NOTE: Also accept if the student just writes yes]		1
Set1,Q3			1
Set2,Q2	A B Y	$\frac{1}{2} + \frac{1}{2}$	
Set3,Q1	0 0 1	/2 1 /2	
2000, 2	0 1 1		
	1 0 1		
	1 1 0		
			1
Set1,Q4			
Set2,Q3			
Set3,Q2	$X_c \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	1	
	ω		
	w		1
Set1,Q5	In amplitude modulation, the amplitude, of the carrier wave, changes in	1	-
Set2,Q4	accordance with the modulating signal, while in frequency modulation,		
Set3,Q3	frequency of the carrier wave varies in accordance with the modulating		1
_	signal.		
	[NOTE: Also accept if the student draws graphs for the two types of		
	modulation]		
~ 1 O 1	Section B	1	
Set 1, Q6	Definition of electric flux ½		
Set2,Q10	S.I. unit		
Set3,Q9	Calculation of flux		
	The 'electric flux', through an elemental area $d\vec{s}$, equals the dot product of	1/2	
	$d\vec{s}$, with the electric field, \vec{E} .	/2	
	[Alternatively: Electric flux is the number of electric field lines passing		
	through a given area.]		
	[Also accept, $d\phi = \vec{E} \cdot d\vec{s}$ Or $\phi = \oint_{s} \vec{E} \cdot d\vec{s}$]		
	[I 130 accept, $u\psi = L$ 13 of $\psi = y_s$ L 13]		
	$(N-m^2)$		
	S.I. units: $\left(\frac{N-m^2}{C}\right)$ or $(V-m)$	1/2	
	$\phi = \vec{E} \cdot \vec{S} = ES(as \ \theta = 0^{o})$		
		1/2	

Page 1 of 17 Final draft 20/07/15 11:00 a.m.



Page 2 of 17

Final draft

20/07/15 11:00 a.m.

Net force on the loop = 5.33×10^7 N The force is directed towards the infinitely long straight wire. OR Formula			1	1
The force is directed towards the infinitely long straight wire. OR Formula Calculation of angle between $\overline{\mu_m}$ and \overline{B} Calculation of $ \overline{\mu_m} $ and torque Torque $= \overline{\mu_m} \times \overline{B}$ $ \overline{\mu_m} = nI \times A = 200 \times 5 \times 100 \times 10^{-4} A \cdot m^2$ $= 10 A \cdot m^2$ Angle between $\overline{\mu_m}$ and $\overline{B} = 90^\circ - 60^\circ = 30^\circ$ $\therefore Torque = 10 \times 0.2 \times \sin 30^\circ$ $= 1 \text{ N-m}$ Naming of the three waves $\frac{1/2}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ $\frac{1}{2} \times \frac{1}{2} + \frac{1}{2}$		Net force on sides bc and da = zero.		
Formula Calculation of angle between $\overline{\mu_m}$ and \overline{B} Calculation of $ \overline{\mu_m} $ and torque Torque = $\overline{\mu_m} \times \overline{B}$ $ \overline{\mu_m} = nI \times A = 200 \times 5 \times 100 \times 10^{-4} A \cdot m^2$ $= 10 A \cdot m^2$ Angle between $\overline{\mu_m}$ and $\overline{B} = 90^\circ - 60^\circ = 30^\circ$ $\therefore Torque = 10 \times 0.2 \times \sin 30^\circ$ $= 1 \text{ N-m}$ Set1,Q9 Set2,Q10 Naming of the three waves Method of production (any one) i. yrays (or X-rays) ii. Ultraviolet rays iii. Infrared rays Production Y rays: (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q10 (a) Finding the transition (b) Radiation of maximum wavelength Justification 2 Set1,Q10 (a) For hydrogen atom, $E_1 = -13.6 \text{ eV}: E_2 = -3.4 \text{ eV}: E_3 = -1.51 \text{ eV}: E_4 = -0.85 \text{ eV}$ $h = 6.63 \times 10^{-34} \times 3 \times 10^0$ $= 496 \times 10^{-9} \times 1.6 \times 10^{-19} \text{ eV}$ $\approx 2.5 \text{ eV}$ This equals (nearly) the difference (E4-E2). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition $= -4$ to $= -3$ corresponds to emission of radiation of maximum wavelength. It is calculation to the properties of the photon of least energy.		\therefore Net force on the loop = 5.33×10^{-7} N		
Formula Calculation of angle between $\overline{\mu_m}$ and \overline{B} $\frac{V_2}{V_2}$ Calculation of $ \overline{\mu_m} $ and torque $ \overline{\mu_m} \times \overline{B} $ $ \overline{\mu_m} = nI \times A = 200 \times 5 \times 100 \times 10^{-4}A - m^2 = 10 A - m^2$ Angle between $\overline{\mu_m}$ and $\overline{B} = 90^\circ - 60^\circ = 30^\circ$ $ \overline{V}_2 $		The force is directed towards the infinitely long straight wire.	1/2	2
Calculation of angle between $\overline{\mu_m}$ and \overline{B} V_2 Calculation of $ \overline{\mu_m} $ and torque $V_2 + V_2$ Torque $= \overline{\mu_m} \times \overline{B}$ $ \overline{\mu_m} = nI \times A = 200 \times 5 \times 100 \times 10^{-4} A - m^2$ $= 10 A - m^2$ V_2 Angle between $\overline{\mu_m}$ and $\overline{B} = 90^{\circ} - 60^{\circ} = 30^{\circ}$ $ V_2 $ Angle between $\overline{\mu_m}$ and $\overline{B} = 90^{\circ} - 60^{\circ} = 30^{\circ}$ $ V_2 $ $= 1 \text{ N-m}$ $ V_2 $ Set1,Q10 Set2,Q10 Naming of the three waves $ V_2 + V_2 + V_2 $ $ $				
Calculation of angle between $\overline{\mu_m}$ and \overline{B} V_2 Calculation of $ \overline{\mu_m} $ and torque $V_2 + V_2$ Torque $= \overline{\mu_m} \times \overline{B}$ $ \overline{\mu_m} = nI \times A = 200 \times 5 \times 100 \times 10^{-4} A - m^2$ $= 10 A - m^2$ V_2 Angle between $\overline{\mu_m}$ and $\overline{B} = 90^{\circ} - 60^{\circ} = 30^{\circ}$ $ V_2 $ Angle between $\overline{\mu_m}$ and $\overline{B} = 90^{\circ} - 60^{\circ} = 30^{\circ}$ $ V_2 $ $= 1 \text{ N-m}$ $ V_2 $ Set1,Q10 Set2,Q10 Naming of the three waves $ V_2 + V_2 + V_2 $ $ $		Formula ½		
Calculation of $ \overline{\mu_m} $ and torque $\frac{1}{2} + \frac{1}{2}$ Torque $= \overline{\mu_m} \times \overline{B}$ $ \overline{\mu_m} = nI \times A = 200 \times 5 \times 100 \times 10^{-4}A - m^2$ $= 10 A - m^2$ Angle between $\overline{\mu_m}$ and $\overline{B} = 90^{\circ} - 60^{\circ} = 30^{\circ}$ $\therefore Torque = 10 \times 0.2 \times \sin 30^{\circ}$ $= 1 \text{ N-m}$ V2 Set1,Q9 Set2,Q10 Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ Method of production (any one) $\frac{1}{2}$ ii. Ultraviolet rays iii. Infrared rays Production $\frac{1}{2} \times \frac{1}{2} \times$				
Torque = $\overline{\mu_m} \times \overline{B}$ $\overline{\mu_m} = nI \times A = 200 \times 5 \times 100 \times 10^{-4} A - m^2$ $\frac{1}{2}$ $\frac{1}{2$				
$ \overrightarrow{\mu_m} = nl \times A = 200 \times 5 \times 100 \times 10^{-4} A - m^2$ $= 10 A - m^2$ Angle between $\overrightarrow{\mu_m}$ and $\overrightarrow{B} = 90^\circ - 60^\circ = 30^\circ$ $\therefore Torque = 10 \times 0.2 \times \sin 30^\circ$ $= 1 \text{ N-m}$ $ Y_2 = 2$ Set1,Q9 Set2,Q10 Set3,Q7 Naming of the three waves Method of production (any one) $ Y_2 = 2$ $ Y_2 = 2$ $ Y_2 = 2$ Set1,Q9 Set3,Q7 Naming of the three waves Method of production (any one) $ Y_2 = 2$ $ Y_2 = 2$ $ Y_2 = 2$ $ Y_2 = 2$ Set1,Q10 Set2,Q9 Set3,Q8 X_1 = 2 \times (10^\circ 10^\circ 10^\circ 10^\circ 10^\circ 10^\circ 10^\circ 10^\circ		Calculation of μ_m and torque		
$ \overrightarrow{\mu_m} = nl \times A = 200 \times 5 \times 100 \times 10^{-4} A - m^2$ $= 10 A - m^2$ Angle between $\overrightarrow{\mu_m}$ and $\overrightarrow{B} = 90^\circ - 60^\circ = 30^\circ$ $\therefore Torque = 10 \times 0.2 \times \sin 30^\circ$ $= 1 \text{ N-m}$ $ Y_2 = 2$ Set1,Q9 Set2,Q10 Set3,Q7 Naming of the three waves Method of production (any one) $ Y_2 = 2$ $ Y_2 = 2$ $ Y_2 = 2$ Set1,Q9 Set3,Q7 Naming of the three waves Method of production (any one) $ Y_2 = 2$ $ Y_2 = 2$ $ Y_2 = 2$ $ Y_2 = 2$ Set1,Q10 Set2,Q9 Set3,Q8 X_1 = 2 \times (10^\circ 10^\circ 10^\circ 10^\circ 10^\circ 10^\circ 10^\circ 10^\circ		Torque = $\overrightarrow{\mu_m} \times \overrightarrow{B}$	1/2	
$= 10 \text{ A-m}^2$ Angle between $\overline{\mu_m}$ and $\overline{B} = 90^{\circ} - 60^{\circ} = 30^{\circ}$ $\therefore Torque = 10 \times 0.2 \times \sin 30^{\circ}$ $= 1 \text{ N-m}$ 2 Set1,Q9 Set2,Q10 Set3,Q7 Naming of the three waves Method of production (any one) i. $y \text{ rays}$ (or X-rays) ii. Ultraviolet rays iii. Infrared rays Production $y \text{ rays}$: (radioactive decay of nuclei) X-rays: (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) (a) Finding the transition (b) Radiation of maximum wavelength $\frac{1}{2}$ $\frac{1}{2}$ $\frac{6.63 \times 10^{-34} \times 3 \times 10^{8}}{1.5 \times 10^{-19} \text{ eV}}$ $\frac{2}{2} \times 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength tip representation is one of least energy.				
Angle between $\overline{\mu_m}$ and $\overrightarrow{B} = 90^{\circ} - 60^{\circ} = 30^{\circ}$ $\therefore Torque = 10 \times 0.2 \times \sin 30^{\circ}$ $= 1 \text{ N-m}$ Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ Set1,Q9 Set2,Q10 Set3,Q7 Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ i. $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ ii. $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ ii. $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ ii. $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ ii. $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$		$= 10 A-m^2$	1/2	
Angle between μ_m and $B = 90^\circ - 60^\circ = 30^\circ$ $\therefore Torque = 10 \times 0.2 \times \sin 30^\circ$ $= 1 \text{ N-m}$ Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ Set 2, Q10 Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ i. \forall rays (or X-rays) $\frac{1}{2}$ $\frac{1}$				
Set1,Q9 Set2,Q10 Set3,Q7 Naming of the three waves Method of production (any one) i. γ rays (or X-rays) ii. Ultraviolet rays iii. Infrared rays Production γ rays: (radioactive decay of nuclei) X-rays: (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength $\frac{1}{2}$ Justification (a) For hydrogen atom, E_1 = -13.6 eV; E_2 = -3.4 eV; E_3 = -1.51 eV; E_4 = -0.85eV h =6.63×10 ⁻³⁴ x 3 × 10 ⁸ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}}$ eV \cong 2.5 eV This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition for adiation of maximum wavelength. Li is expecuse this transmits on gives out the photon of least energy.		Angle between \overrightarrow{u} and $\overrightarrow{R} = 90^{\circ} - 60^{\circ} = 30^{\circ}$	1/2	
Set1,Q9 Set2,Q10 Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ Method of production (any one) $\frac{1}{2}$ $\frac{1}{2}$ $\frac{1}{2$				
Set1,Q9 Set2,Q10 Set3,Q7 Naming of the three waves Method of production (any one) i. γ rays (or X-rays) ii. Ultraviolet rays iii. Infrared rays Production γ rays : (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification 1 (a) For hydrogen atom, $E_1 = -13.6 \text{ eV}$; $E_2 = -3.4 \text{ eV}$; $E_3 = -1.51 \text{ eV}$; $E_4 = -0.85 \text{ eV}$ $h = 6.63 \times 10^{-34} \times 3 \times 10^8$ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is a begause this transmission gives out the photon of least energy.		$ 101 que = 10 \times 0.2 \times \sin 30 $		
Set1,Q9 Set2,Q10 Set3,Q7 Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} = \frac{1}{2}$ i. $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} = \frac{1}{2}$ ii. $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} = \frac{1}{2}$ iii. Ultraviolet rays iii. Infrared rays Production $\frac{1}{2} + \frac{1}{2} = \frac{1}{2}$ Production $\frac{1}{2} + \frac{1}{2} = \frac{1}{2}$ Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2} = \frac{1}{2}$ ii. $\frac{1}{2} + \frac{1}{2} = \frac{1}{2}$		=1 N-m	1/2	2
Set2,Q10 Set3,Q7 Naming of the three waves Method of production (any one) i. γ rays (or X-rays) ii. Ultraviolet rays iii. Infrared rays Production γ rays: (radioactive decay of nuclei) X-rays: (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1 = -13.6 \text{ eV}$; $E_2 = -3.4 \text{ eV}$; $E_3 = -1.51 \text{ eV}$; $E_4 = -0.85 \text{ eV}$ $h = 6.63 \times 10^{-34} \text{ y } 3 \times 10^8$ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{4.96 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV}$ $\approx 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is expective this transmission gives out the photon of least energy.	Set1.O9			
Set3,Q7 Method of production (any one) ½		Naming of the three waves $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$		
i. γ rays (or X-rays) ii. Ultraviolet rays iii. Infrared rays Production γ rays: (radioactive decay of nuclei) X-rays: (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1=-13.6 \text{ eV}: E_2=-3.4 \text{ eV}; E_3=-1.51 \text{ eV}; E_4=-0.85 \text{ eV}$ $h=6.63\times10^{-34}\text{ ys}; c=3\times10^8\text{ms}^{-1}$ Photon Energy = $\frac{h}{2}$ = $\frac{6.63\times10^{-34}\times3\times10^8}{496\times10^{-9}\times1.6\times10^{-19}}$ eV $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is a beginner that the required transition gives out the photon of least energy.		Method of production (any one) ½		
ii. Ultraviolet rays iii. Infrared rays Production γ rays: (radioactive decay of nuclei) X-rays: (X-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) 1/2 2	2010, Q			
ii. Ultraviolet rays iii. Infrared rays Production γ rays: (radioactive decay of nuclei) X-rays: (X-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) 1/2 2		i. v rays (or X-rays)	1/2	
iii. Infrared rays Production γ rays: (radioactive decay of nuclei) X-rays: (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1 = -13.6 \text{ eV}$; $E_2 = -3.4 \text{ eV}$; $E_3 = -1.51 \text{ eV}$; $E_4 = -0.85 \text{ eV}$ $h = 6.63 \times 10^{-34} \text{ Js}$; $c = 3 \times 10^8 \text{ ms}^{-1}$ Photon Energy= $\frac{hc}{\lambda}$ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E_4 - E_2). Hence the required transition is ($n = 4$) to ($n = 2$) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition $n = 4$ to $n = 3$ corresponds to emission of radiation of maximum wavelength. It is a because this transmission gives out the photon of least energy.				
Production γ rays: (radioactive decay of nuclei) X -rays: (x -ray tubes or inner shell electrons) UV - rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one)		· · · · · · · · · · · · · · · · · · ·		
$\begin{array}{c} \gamma \ rays: (radioactive decay of nuclei) \\ X-rays: (x-ray tubes or inner shell electrons) \\ UV-\ rays: (Movement from one inner energy level to another) \\ Infrared rays: (vibration of atoms and molecules) \\ (Any one) \\ & $			/ 2	
X-rays : (x-ray tubes or inner shell electrons) UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1 = -13.6 \text{ eV}; E_2 = -3.4 \text{ eV}; E_3 = -1.51 \text{ eV}; E_4 = -0.85 \text{ eV}$ $h = 6.63 \times 10^{-34} \text{ Js}; c = 3 \times 10^{8} \text{ ms}^{-1}$ Photon Energy= $\frac{hc}{\lambda}$ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^{8}}{496 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV}$ $\approx 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy.				
UV- rays: (Movement from one inner energy level to another) Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition 1 1 (b) Radiation of maximum wavelength $\frac{1}{2}$ Justification $\frac{1}{2}$ (a) For hydrogen atom, E_1 = -13.6 eV; E_2 = -3.4 eV; E_3 = -1.51 eV; E_4 = -0.85eV E_2 = -3.4 eV; E_3 = -1.51 eV; E_4 = -0.85eV E_3 = -1.51 eV; E_3 = -1.51 eV; E_4 = -0.85eV E_3 = -1.51 eV; E_3 = -1.51 eV; E_4 = -0.85eV E_3 = -1.51 eV; E				
Infrared rays: (vibration of atoms and molecules) (Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1 = -13.6 \text{ eV}$; $E_2 = -3.4 \text{ eV}$; $E_3 = -1.51 \text{ eV}$; $E_4 = -0.85 \text{ eV}$ $h = 6.63 \times 10^{-34} \text{ Js}$; $c = 3 \times 10^8 \text{ ms}^{-1}$ Photon Energy = $\frac{hc}{\lambda}$ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^3}{496 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV}$ $\approx 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy				
(Any one) Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1=-13.6 \text{ eV}$; $E_2=-3.4 \text{ eV}$; $E_3=-1.51 \text{ eV}$; $E_4=-0.85 \text{ eV}$ $h=6.63\times10^{-34} \text{ Js}$; $c=3\times10^8 \text{ms}^{-1}$ Photon Energy= $\frac{hc}{\lambda}$ $= \frac{6.63\times10^{-34}\times3\times10^3}{496\times10^{-9}\times1.6\times10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy				
Set1,Q10 Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength $\frac{1}{2}$ Justification $\frac{1}{2}$ (a) For hydrogen atom, $E_1=-13.6 \text{ eV}$; $E_2=-3.4 \text{ eV}$; $E_3=-1.51 \text{ eV}$; $E_4=-0.85 \text{ eV}$			1/2	
Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1=-13.6 \text{ eV}$; $E_2=-3.4 \text{ eV}$; $E_3=-1.51 \text{ eV}$; $E_4=-0.85 \text{ eV}$ $h=6.63\times 10^{-34} \text{ Js}$; $c=3\times 10^8 \text{ ms}^{-1}$ Photon Energy= $\frac{hc}{\lambda}$ $=\frac{6.63\times 10^{-34}\times 3\times 10^8}{496\times 10^{-9}\times 1.6\times 10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy			/-	2
Set2,Q9 Set3,Q8 (a) Finding the transition (b) Radiation of maximum wavelength Justification (a) For hydrogen atom, $E_1=-13.6 \text{ eV}$; $E_2=-3.4 \text{ eV}$; $E_3=-1.51 \text{ eV}$; $E_4=-0.85 \text{ eV}$ $h=6.63\times 10^{-34} \text{ Js}$; $c=3\times 10^8 \text{ ms}^{-1}$ Photon Energy= $\frac{hc}{\lambda}$ $=\frac{6.63\times 10^{-34}\times 3\times 10^8}{496\times 10^{-9}\times 1.6\times 10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy	Set1,Q10			
Set3,Q8 (a) For hydrogen atom, $E_1 = -13.6 \text{ eV}; E_2 = -3.4 \text{ eV}; E_3 = -1.51 \text{ eV}; E_4 = -0.85 \text{ eV}$ $h = 6.63 \times 10^{-34} \text{ Js}; c = 3 \times 10^8 \text{ ms}^{-1}$ Photon Energy = $\frac{hc}{\lambda}$ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy.		(a) Finding the transition		
Justification (a) For hydrogen atom, $E_{1}=-13.6 \text{ eV}; E_{2}=-3.4 \text{ eV}; E_{3}=-1.51 \text{ eV}; E_{4}=-0.85 \text{ eV}$ $h=6.63\times10^{-34} \text{ Js}; c=3\times10^{8} \text{ms}^{-1}$ Photon Energy= $\frac{hc}{\lambda}$ $=\frac{6.63\times10^{-34}\times3\times10^{8}}{496\times10^{-9}\times1.6\times10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy.		(b) Radiation of maximum wavelength ½		
E ₁ = -13.6 eV; E ₂ = -3.4 eV; E ₃ = -1.51 eV; E ₄ = -0.85eV h=6.63×10 ⁻³⁴ Js; c=3×10 ⁸ ms ⁻¹ Photon Energy= $\frac{hc}{\lambda}$ = $\frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}}$ eV $\cong 2.5$ eV This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength.	, (Justification ½		
E ₁ = -13.6 eV; E ₂ = -3.4 eV; E ₃ = -1.51 eV; E ₄ = -0.85eV h=6.63×10 ⁻³⁴ Js; c=3×10 ⁸ ms ⁻¹ Photon Energy= $\frac{hc}{\lambda}$ = $\frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}}$ eV $\cong 2.5$ eV This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength.				
E ₁ = -13.6 eV; E ₂ = -3.4 eV; E ₃ = -1.51 eV; E ₄ = -0.85eV h=6.63×10 ⁻³⁴ Js; c=3×10 ⁸ ms ⁻¹ Photon Energy= $\frac{hc}{\lambda}$ = $\frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}}$ eV $\cong 2.5$ eV This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength.		(a) For hydrogen atom,		
h=6.63×10 ⁻³⁴ Js; c=3×10 ⁸ ms ⁻¹ Photon Energy= $\frac{hc}{\lambda}$ = $\frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}}$ eV $\cong 2.5$ eV This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy		$E_1 = -13.6 \text{ eV}$; $E_2 = -3.4 \text{ eV}$; $E_3 = -1.51 \text{ eV}$; $E_4 = -0.85 \text{eV}$		
Photon Energy= $\frac{hc}{\lambda}$ $= \frac{6.63 \times 10^{-34} \times 3 \times 10^{8}}{496 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV}$ $\cong 2.5 \text{ eV}$ This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy		$h=6.63\times10^{-34} \text{ Js}; c=3\times10^8 \text{ms}^{-1}$		
=\frac{6.63 \times 10^{-34} \times 3 \times 10^8}{496 \times 10^{-9} \times 1.6 \times 10^{-19}} \text{ eV} \times 2.5 \text{ eV} This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy				
 ≅ 2.5 eV This equals (nearly) the difference (E₄-E₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy. 			1/2	
 ≅ 2.5 eV This equals (nearly) the difference (E₄-E₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy. 		$=\frac{6.63 \times 10^{-31} \times 3 \times 10^{3}}{6} \text{ eV}$		
This equals (nearly) the difference (E ₄ -E ₂). Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy				
Hence the required transition is (n=4) to (n=2) [Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy				
[Alternatively: If the candidate calculates by using Rydberg formula, and identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy				
identifies correctly the required transition, full credit may be given.] (b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy			1/2	
(b) The transition n=4 to n=3 corresponds to emission of radiation of maximum wavelength. It is so because this transmission gives out the photon of least energy		, , , ,		
maximum wavelength. It is so because this transmission gives out the photon of least energy				
It is so because this transmission gives out the photon of least energy		<u> </u>	1/2	
It is so because this transmission gives out the photon of least energy. $\frac{1}{2}$		<u> </u>		
		It is so because this transmission gives out the photon of least energy.	1/2	2

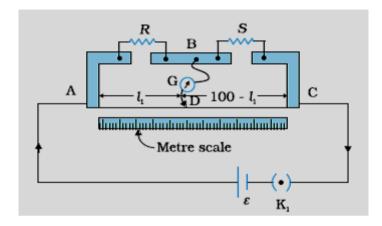
Page 3 of 17 Final draft 20/07/15 11:00 a.m.

0 1 011	Section C		
Set1,Q11 Set2,Q19 Set3,Q16	(a) Derivation of the relation between I and $ \overrightarrow{v_d} $ 2 (b) Calculation of the charge flowing in 10 s 1		
	(a) According to the figure, $\Delta x = v_d st$ Hence, amount of charge crossing area A in time Δt	1/2	
	$\Delta x = v_d \Delta t$	1/2	
	$\therefore \Delta Q = I\Delta t = neA v_d \Delta t$	1/2	
	$\therefore I = neAv_d$	1/2	
	(b) Charge flowing = $\sum I \Delta t$ =area under the curve	1/2	
	$= \left[\frac{1}{2} \times 5 \times 5 + 5(10 - 5)\right] C$ = 37.5 C	1/2	3
Set1,Q12 Set2,Q20 Set3,Q17	Circuit Diagram Working Principle Derivation of necessary formula The circuit diagram, of the potentiometer, is as shown here:	1	
	Working Principle: The potential drop, V , across a length l of a uniform wire, is proportional to the length l of the wire. (or $V \propto l$ (for a uniform wire) Derivation: Let the points 1 and 3 be connected together. Let the balance point be at the point N_1 where $AN_1=l_1$ Next let the points 2 and 3 be connected together. Let the balance point be at the point N_2 where $AN_2=l_2$. We then have	1/2	
	$arepsilon_1 = k l_1$ and $arepsilon_2 = k l_2$	1/2	
	ge 4 of 17 Final draft 20/07/1	 5 11:00 a	

Page 4 of 17 Final draft 20/07/15 11:00 a.m.

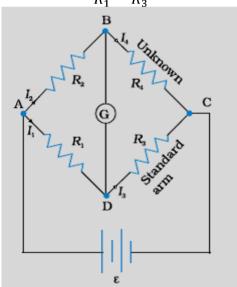
Circuit Diagram	1/2
Working Principle	1
Determination of unknown resistance	1
Precautions	1/2

The circuit diagram of the meter bridge is as shown below:



Working Principle: The working principle of the meter bridge is the same as that of a wheatstone bridge. The Wheatstone bridge gets balanced when:

$$\frac{R_2}{R_1} = \frac{R_4}{R_3}$$



For the meter bridge, circuit shown above, this relation takes the form

$$\frac{R}{S} = \frac{l_1}{(100 - l_1)}$$

Determination of unknown Resistance (*R*):

In the circuit diagram shown above, S is taken as a known standard resistance. We find the value of the balacing length l_1 , corresponding to a given value of *S*. We then use the relation:

Page 5 of 17 Final draft 20/07/15 11:00 a.m.

1/2

1/2

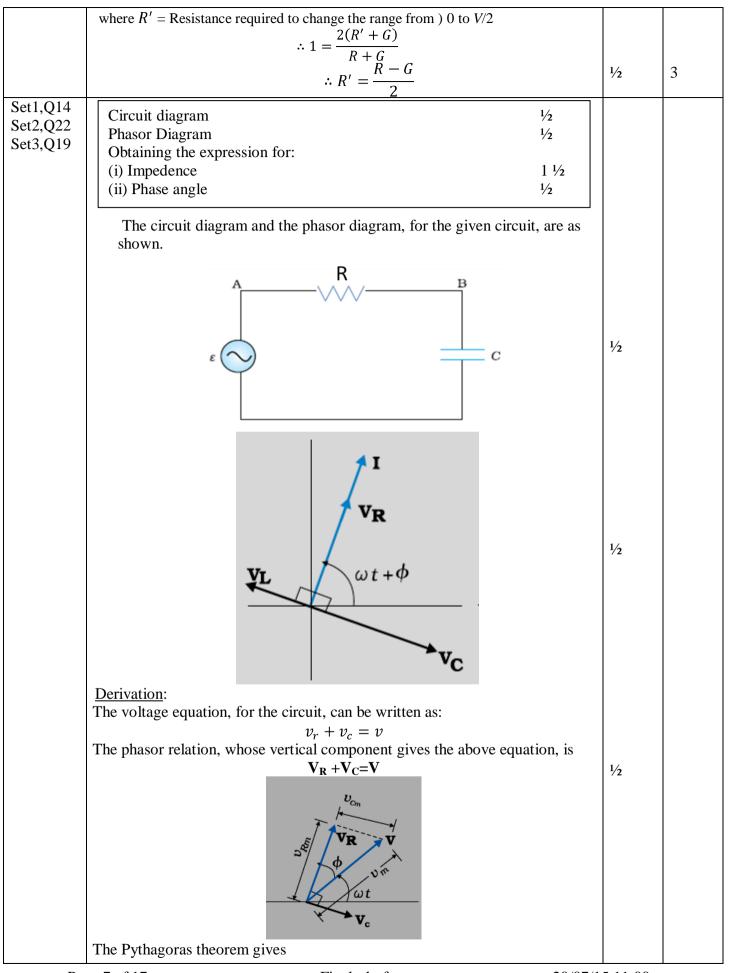
 $\frac{1}{2}$

 $\frac{1}{2}$

1/2

	R l_1		
	$\frac{R}{S} = \frac{l_1}{(100 - l_1)}$		
	to calculate <i>R</i> . By choosing (at least three) different value of <i>S</i> , we calculate <i>R</i> each time. The average of these values of <i>R</i> gives the value of the unknown resistance.	1/2	
	Precautions: (1) Make all contacts in a neat, clean and tight manner (2) Select those values of <i>S</i> for which the balancing length is close to the middle point of the wire. [Any one]	1/2	3
Set1,Q13 Set2,Q21 Set3,Q18	(a) Need for having a radial Magnetic field $\frac{1}{Achieving}$ the radial field $\frac{1}{V_2}$ (b) Formula $\frac{1}{V_2}$ Calculation of the required resistance $\frac{1}{Achieving}$ The relation between the current (i) flowing through the galvanometer coil, and the angular deflection (ϕ) of the coil (from its equilibrium position), is $\phi = \left(\frac{NABI}{k}\right)$ where θ is the angle between the magnetic field \vec{B} and the equivalent magnetic moment $\vec{\mu}_m$ of the current carrying coil. Thus I is not directly proportional to ϕ . We can ensure this proportionality by having $\theta = 90^\circ$. This is possible only when the magnetic field, \vec{B} , is a radial magnetic field. In such a field, the plane of the rotating coil is always parallel to \vec{B} . To get a radial magnetic field, the pole pieces of the magnet, are made concave in shape. Also a soft iron cylinder is used as the core. [Alternatively: Accept if the candidate draws the following diagram to achieve the radial magnetic field.]	1/2 1/2 1/2	
	$I_m = \frac{\left(\frac{V}{2}\right)}{R' + G}$		
_	ge 6 of 17 Final draft 20/07/1	5 11.00 a	

Page 6 of 17 Final draft 20/07/15 11:00 a.m.



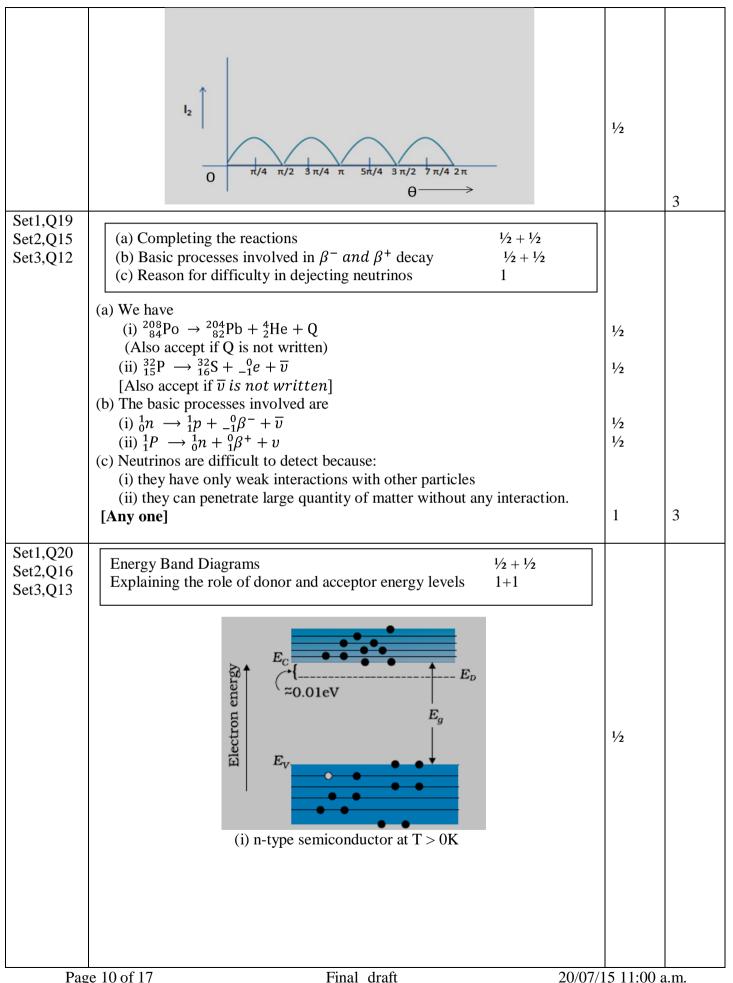
Page 7 of 17 Final draft 20/07/15 11:00 a.m.

		ı	
Su	$v_m^2 = v_{RM}^2 + v_{cm}^2$ ubstituting the values of v_{RM} and v_{cm} , into this equation, gives	1/2	
	$v_m^2 = (i_m R)^2 + (i_m X_C)^2 = i_m^2 (R^2 + X_C^2)$		
	$i_m = \frac{v_m}{\sqrt{R^2 + X_c^2}}$	1/	
	$\sqrt{R^2 + X_c^2}$ The impedance of the circuit is given by:	1/2	
	$7 - \sqrt{p^2 + v^2} - \sqrt{p^2 + \frac{1}{2}}$		
	$\mathbf{Z} = \sqrt{R^2 + X_c^2} = \sqrt{R^2 + \frac{1}{\omega^2 c^2}}$ The phase angle ϕ is the angle between V_R and V . Hence		
	$\tan \phi = \frac{X_C}{R} = \frac{1}{\omega CR}$	1/2	3
Set1,Q15	(i) Formula for magnetic moment ½		3
Set 2,Q11 Set 3,O20	Calculation of magnetic moment (ii) Formula for torque 72 1 (ii) Formula for torque		
	Calculation of torque 1		
$\mid \mu_r \mid$	Associated magnetic moment $m = niA$	1/2 1/2	
	$= 2000 \times 4 \times 1.6 \times 10^{-4} \text{ A} - \text{m}^2$ = 1.28 A - \text{m}^2	1/2	
	i) torque $=\mu_m B \sin \theta$	1/2	
	$1.28 \times 7.5 \times 10^{-2} \times \sin 30^{o}$ = 0.048 N - m	1/2 1/2	3
Cat 1 O 1 6			
/	(a) Formula ½ Coloulation of the ratio		
Set3,Q21	Calculation of the ratio (b) Answering about Conservation of Energy Explanation 1 1 1 1 1 1 1 1 1 1 1 1 1		
		1/	
(a	$a) \frac{I_{max}}{I_{min}} = \left \frac{a_1 + a_2}{a_1 - a_2} \right ^2$	1/2	
H	Here $\frac{a_1}{a_2} = \sqrt{\frac{W_2}{W_1}} = \sqrt{\frac{4}{1}} = \frac{2}{1}$	1/2	
	$\frac{I_{max}}{I_{min}} = \left \frac{2a_2 + a_2}{2a_2 - a_2} \right ^2 = 9:1$	1/2	
(b	b) There is NO violation of the conservation of energy.	1/2	
TI	he appearance of the bright and dark fringes is simply due to a edistribution of energy'.	1	3
Set1,Q17 Set2,Q13	(a) Factors by which the resolving power can be increased. 1		-
Set3,Q22	(b) Formula 1/2 Estimation of angular separation 1 1/2		
(a	1) The resolving power of a telescope can be increased by		

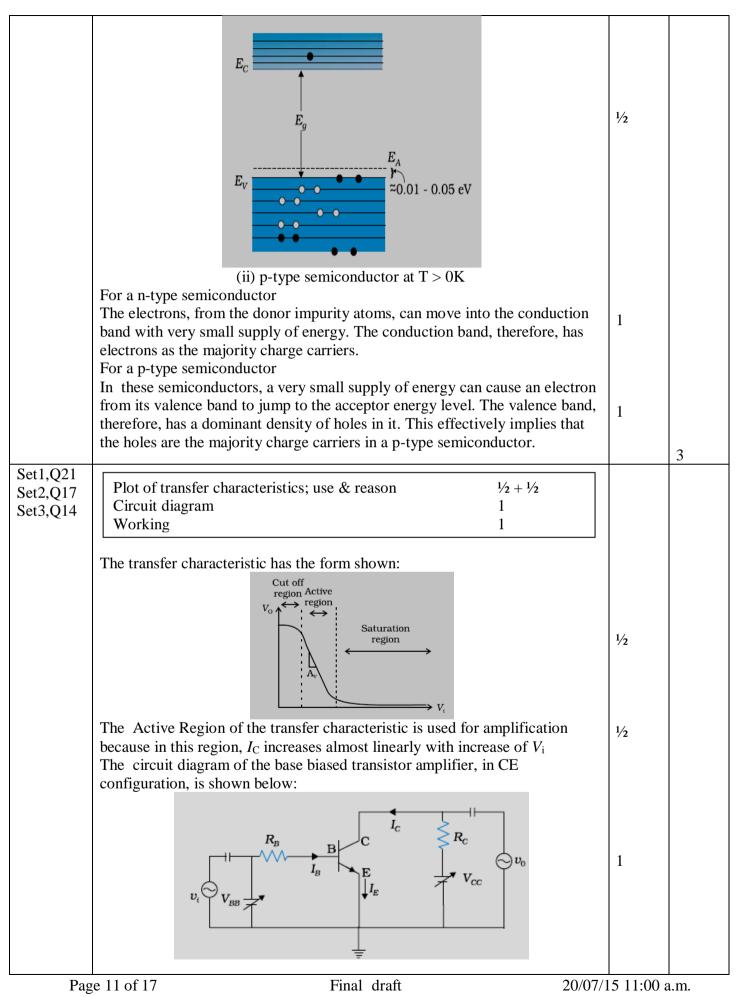
Page 8 of 17 Final draft 20/07/15 11:00 a.m.

	(i) increasing the diameter of its objective		
	(ii) using light of short wavelength		
	[Note: Give full credit even if a student writes just the first of these two factors.]	1	
	(b) Position of Maxima: $\theta \approx \left(n + \frac{1}{2}\right) \frac{\lambda}{a}$; position of minima = $\frac{n\lambda}{a}$	1/2	
	For first order maxima, $\theta = \frac{3\lambda}{2a}$	1/2	
	and for third order minima, $\theta = \frac{3\lambda}{a}$		
	∴ Required angular separation $3\lambda = 3 \times 600 \times 10^{-9}$	1/2	
	$= \frac{3\lambda}{2a} = \frac{3 \times 600 \times 10^{-9}}{2 \times 1 \times 10^{-3}} $ radian = 9 × 10 ⁻⁴ radian	1/2	3
Set1,Q18 Set2,Q14 Set3,Q11	(a) Reason for preferring sun glasses made up of polaroids (b) Formula for intensity of light transmitted through P_2 Plot of I vs θ 1 1/2		
	(a) Polaroid sunglasses are preferred because they can be much more effective than coloured sunglasses in cutting off the harmful (UV) rays of the sun.	1	
	[Alternatively: Poloroid sun glasses are prefered over coloured sun glasses because they are more effective in reducing the glare due to reflections from horizontal surfaces.] [Alternatively: Poloroid sun glasses are prefered over coloured sun glasses because they provide a better protection to our eyes.]		
	$\begin{array}{c} P_1 \\ \hline \\ P_2 \\ \hline \\ I_1 \end{array}$		
	Let θ be the angle between the pass axis of P_1 and P_3 . The angle between the pass axis of P_3 and P_2 would then be $\left(\frac{\pi}{2} - \theta\right)$. By Malus' law,	1/2	
	$I_3 = I_1 cos^2 \theta$	1/2	
	and $I_2 = I_3 \cos^2\left(\frac{\pi}{2} - \theta\right) = I_3 \sin^2\theta$		
	$\therefore I_2 = I_1 \cos^2 \theta \sin^2 \theta = \frac{I_1 (\sin 2\theta)^2}{4}$	1/2	
	The plot of I_2 vs θ , therefore, has the form shown below:		

Page 9 of 17 Final draft 20/07/15 11:00 a.m.



Page 10 of 17

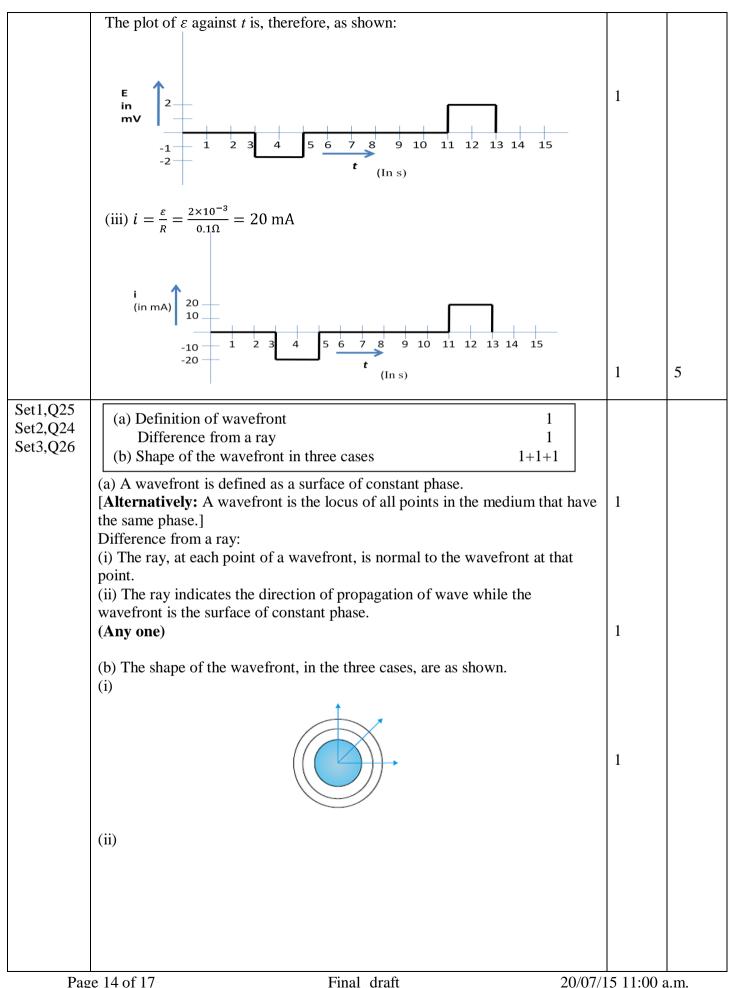


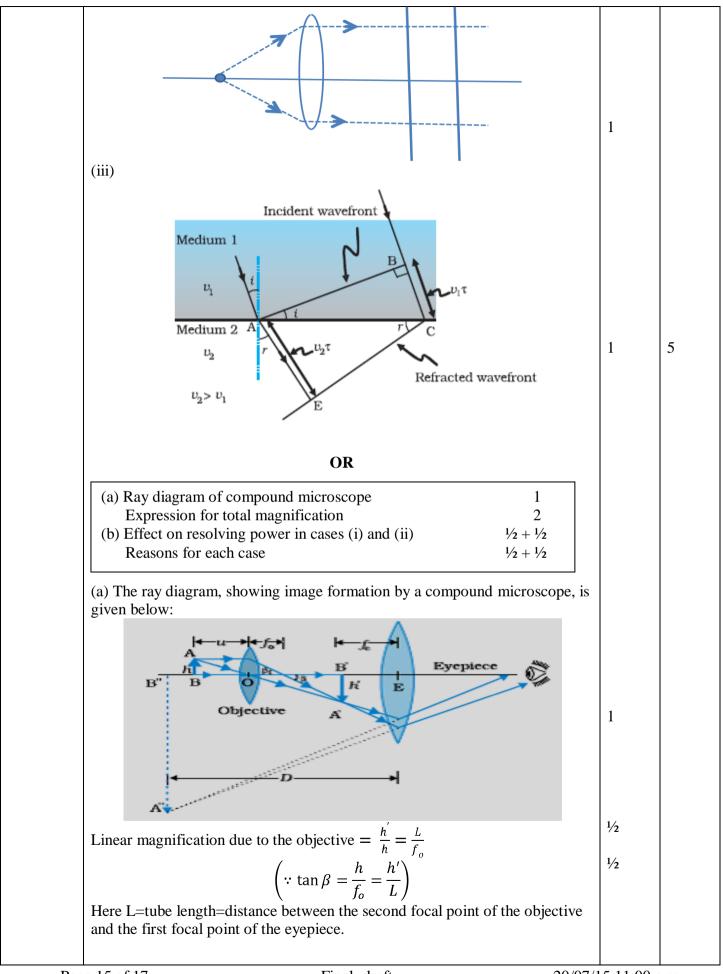
		1	,
Set1,Q22	Working: The sinusoidal voltage, superposed on the dc base bias, causes the base current to have sinusoidal variations. As a result the collector current, also has similar sinusoidal variations present in it. The output, between the collector and the ground, is an amplified version of the input sinusoidal voltage. (Also accept 'other forms' for explanation of 'working'	1	3
Set1,Q22 Set2,Q18	Explanation of each of three terms 1+1+1		
Set3,Q15	 (i) Internet Surfing Visiting, or using, the different websites on the internet. (ii) Social networking Social networking implies using site like (a) Facebook, Twitter, etc, to share ideas and information with a large number of people. (b) Using internet for chatting, video sharing, etc, among friends and acquaintances. (Any one) (iii) E-mail 	1	
	Using internet(rather than the post office) for exchanging (multimedia)	1	
	communication between different persons and organizations.		3
	Section D		
Set1,Q23	Section D		
Set2,Q23 Set3,Q23	(1) Value displayed by Dr. Kapoor Bimla's parents 1+1 (2) Reason for safety 1 (3) Definition and Significance 1/2 + 1/2		
	(1) Dr. Kapoor : Helpful & Considerate Bimla's Parents: Gratefulness	1	
	(2) It is considered safe to be inside a car during lightening and thunderstorm as the electric field inside a conductor is zero.	1	
	(3) Dielectric strength of a dielectric indicates the strength of the electric field that a dielectric can withstand without breaking down.	1/2	
	This signifies the maximum electric field up to which the dielectric can safely play its role.	1/2	
	Section E		4
Set1,Q24			
Set2,Q26 Set3,Q25	(a) Statement of Lenz's law Predicting the polarity (b) (i) Formula Substitution and Calculation (ii) Effect on voltage 1		
	(a) Lenz's law: The polarity of induced emf is such that it tends to produce a current which opposes the change in magnetic flux that produced it.	1	

Page 12 of 17 Final draft 20/07/15 11:00 a.m.

		1
Polarity A \rightarrow (+ve); B \rightarrow (-ve)	1/2 + 1/2	
(b) (i) $V = Bl\theta$,2 1 ,2	
Here B = vertical component of Earth's magnetic field B= $(5\times10^{-4} \sin 30^{\circ})$ T = 2.5×10^{-4} T	1/2	
	1/2	
$\therefore V = \left[2.5 \times 10^{-4} \times 25 \times \frac{1800 \times 10^{3}}{60 \times 60} \right] \text{volt}$		
= 3.125 volt	1/2 1/2	
(ii) Now B = horizontal component of Earth's magnetic field	, 2	
$=B\cos 30^{\circ} = \frac{B\sqrt{3}}{2}$	1/-	
∴ $V' = \sqrt{3}V = 1.732 \times 3.125 \text{ volt} \cong 5.4 \text{ volt}$	1/2 1/2	5
OR		
Definition of mutual inductance 1		
Factors affecting mutual inductance 1		
Formulae for the three cases Calculations for plotting the graphs 1		
Plots of three graphs $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$		
Mutual Inductance: The mutual inductance, of a pair of coils, equals the magnetic flux linked		
with one of them due to a unit current in the other.		
Alternatively, The mutual inductance, of a pair of coils, equals the emf	1	
induced in one of them when the rate of change of current in the other is unity.	1	
Factors affecting the mutual inductance of a pair of coils		
(i) The sizes of the two coils		
(ii) The shape of the two coils		
(iii) the distance of separation between the two coils		
(iv) The nature of the medium between the two coils(v) The relative orientation of the two coils.		
[NOTE: Any two]	1/2 + 1/2	
From $t = 0$ to $t = 3s \left(= \frac{30 \text{ cm}}{10 \text{ cm/s}} \right)$, the flux through the coil is zero.		
From $t = 3$ s to $t = 5$ s, the flux through the coil increases from 0 to		
$\left[0.1 \times \left(\frac{20}{100}\right)^2\right]$ Wb, ie 0.004 Wb.		
From $t = 5$ s to $t = 11$ s, the flux remains constant at the value 0.004 Wb.		
From $t = 11$ s to $t = 13$ s, the flux through the coil remains zero.		
(i) The plot of ϕ against t is, therefore, as shown:		
0.004		
Φ0.004		
	1	
(In Wb)		
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15		
t (in ms)		
(ii) $\varepsilon = -\frac{u\varphi}{dt}$		
Page 13 of 17 Final draft 20/07/	15 11:00 a	a m

Page 13 of 17 Final draft 20/07/15 11:00 a.m.



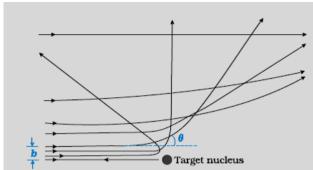


	When the final image is formed at infinity, the angular magnification due to the eye piece equals $\frac{D}{f_e}$. (D=least distance of distinct vision)	1/2	
	$\therefore \text{ Total magnification when the final image is formed at infinity} = \left(\frac{L}{f_o} \cdot \frac{D}{f_e}\right)$	1/2	
	(c) (i) Resolving power increases when the focal length of the objective is decreased.	1/ ₂ 1/ ₂	
	(d) This is because the minimum separation, $d_{min} \left(= \frac{1.22 f \lambda}{D} \right)$ decreases		
	when f is decreased.	1/2	
	(ii) Resolving power decreases when the wavelength of light is increased.	72	
	This is because the minimum separation, $d_{min} \left(= \frac{1.22 f \lambda}{d} \right)$	1/2	5
Set1,Q26	increases when λ is increased.		
Set1,Q26 Set2,Q25 Set3,Q24	(a) Writing three features Explanation on the basis of Einstein's photoelectric equation $\frac{1}{2} + \frac{1}{2} + \frac{1}{2}$ (b) (i) Reason for equality of the two slopes (ii) Identification of material		
	(a) Three features, of photoelectric effect, which cannot be explained by the wave theory of light, are:(i) Maximum kinetic energy of emitted electrons is independent of the	1/2	
	intensity of incident light.		
	 (ii) There exists a 'threshold frequency' for each photosensitive material. (iii) 'Photoelectric effect' is instantaneous in nature. Einstein's photoelectric equation 	1/2 1/2	
	$K_{max} = h\nu - \phi_o$ [Alternatively: $eV_o = h\nu - \phi_o$] can be used to explain these features as follows.		
	(i) Einstein's equation shows that $K_{max} \propto \nu$. However, K_{max} does not depend on the intensity of light.	1/2	
	(ii) Einstein's equation shows that for $\nu < \frac{\phi_0}{h}$, K_{max} becomes negative, i.e,	1/2	
	there cannot be any photoemission for $\nu < \nu_o(\nu_o = \frac{\phi_o}{h})$ (iii) The free electrons in the metal, that absorb completely the energy of the incident photons, get emitted instantaneously.	1/2	
	(b) (i) Slope of the graph between V_o and v (from Einstein's equation) equals (h/e) . Hence it does not depend on the nature of the material.	1	
	(ii) Emitted electrons have greater energy for material M_1 . This is because $\phi_o(=h\nu_o)$ has a lower value for material M_1 .	1	5
	OR		

Page 16 of 17 Final draft 20/07/15 11:00 a.m.

(a) Drawing the Trajectory	1	
Estimating the size of the nucleus	1	
(b) Establishment of wave nature	1	
(c) Estimating the ratio of deBroglie wavelengths	2	

(a) The trajectory, traced by the α –particles in the Coulomb field of target nucleus, has the form shown below.



The size of the nucleus was estimated by observing the distance (d) of closest approach, of the α -particles. This distance is given by:

$$\frac{1}{4\pi\varepsilon_o} \cdot \frac{(Ze)(2e)}{d} = K$$

where K=kinetic energy of the α -particles when they are far away from the target nuclei.

(b) The wave nature of moving electrons was established through the Davisson-Germer experiment.

In this experiment, it was observed that a beam of electrons, when scattered by a nickel target, showed 'maxima' in certain directions; (like the 'maxima' observed in interference/diffraction experiments with light.)

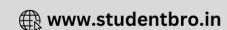
(c) We have:
$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mqV}}$$

$$\therefore \frac{\lambda_d}{\lambda_\alpha} = \sqrt{\frac{m_\alpha q_\alpha}{m_d q_d}}$$

$$= \sqrt{2 \times 2} = 2$$
1/2

Page 17 of 17 Final draft 20/07/15 11:00 a.m.





1

1/2

 $\frac{1}{2}$